

Crystallographic report

(4,7-Diphenyl-1,10-phenanthroline) bis(pyrrolinedithiocarbamato)zinc(II) chloroform solvate

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The mononuclear structure of $\text{Zn}(\text{S}_2\text{CN}(\text{CH}_2)_4)_2(4,7\text{-Ph}_2\text{-1,10-phenanthroline})$ shows the zinc atom in each of the two independent molecules comprising the asymmetric unit to exist in a distorted octahedral geometry defined by an N_2S_4 donor set. Copyright © 2003 John Wiley & Sons, Ltd.

KEYWORDS: crystal structure; zinc; dithiocarbamate; diimine adduct

COMMENT

In connection with ongoing studies into the structural chemistry of diimine adducts of the zinc-triad 1,1-dithiolates,¹⁻³ the title compound, Zn(S₂CN(CH₂)₄)₂(4,7-Ph₂-1,10-phen), was investigated. Two crystallographically independent molecules comprise the asymmetric unit; they are chemically similar but differ in their geometric parameters, most notably in the Zn—S bond distances (Fig. 1). Thus, the Zn—S distances range from 2.4524(8) to 2.5788(8) Å for the Zn1 atom and from 2.4337(8) to 2.5678(8) Å for Zn2. The octahedral geometry found for each of the zinc atoms is defined by an N₂S₄ donor set.

EXPERIMENTAL

Bright-yellow crystals were isolated from an acetonitrile/chloroform (1:1) solution containing equimolar amounts of $\text{Zn}(\text{S}_2\text{CN}(\text{CH}_2)_4)_2$ and 4,7-Ph₂-phen (Aldrich); m.p. 205–208 °C. IR (KBr, Ca^{-1}): $\nu(\text{C—S})$ 1005, 945 and $\nu(\text{C—N})$ 1428. Intensity data were collected at 183 K on a Bruker AXS SMART CCD diffractometer for a yellow block $0.18 \times 0.23 \times 0.49 \text{ mm}^3$. $\text{C}_{34}\text{H}_{32}\text{N}_4\text{S}_4\text{Zn}\cdot\text{CHCl}_3$, $M = 809.6$, triclinic, $P\bar{1}$, $a = 13.1703(6)$, $b = 17.6586(9)$, $c = 17.8809(9) \text{ \AA}$, $\alpha = 63.013(1)$, $\beta = 86.826(1)$, $\gamma = 75.315(1)^\circ$, $V = 3576.1(3) \text{ \AA}^3$, $Z = 4$, 20 489 unique data ($\theta_{\text{max}} = 30.0^\circ$), $R = 0.078$ (all data), $wR = 0.169$ (all data), $\rho_{\text{max}} = 1.34 \text{ e}^{-} \text{ \AA}^{-3}$ (near solvent molecule). Disorder in the structure was noted and resolved for the C8 atom of the pyrrolidine ring. From refinement, the major component had a site occupancy of 0.58. Some

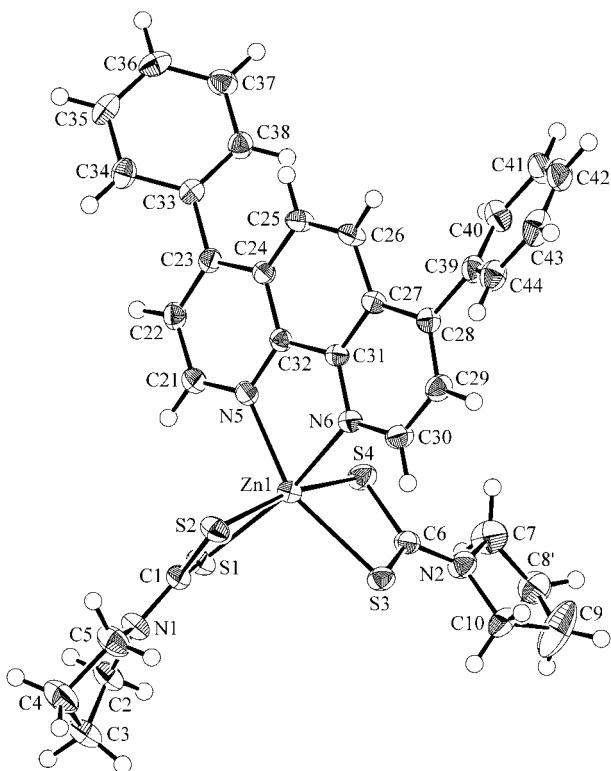


Figure 1. Molecular structure of molecule a in $\text{Zn}(\text{S}_2\text{CN}(\text{CH}_2)_4)_2(4,7\text{-Ph}_2\text{1,10-phenanthroline})$; the chloroform molecule of solvation is omitted. Key geometric parameters: $\text{Zn}(1)—\text{S}(1)$ 2.4524(8), $\text{Zn}(1)—\text{S}(2)$ 2.5051(8), $\text{Zn}(1)—\text{S}(3)$ 2.4993(8), $\text{Zn}(1)—\text{S}(4)$ 2.5788(8), $\text{Zn}(1)—\text{N}(5)$ 2.167(2), $\text{Zn}(1)—\text{N}(6)$ 2.195(2) Å, $\text{S}(1)—\text{Zn}(1)—\text{S}(2)$ 73.22(3), $\text{S}(3)—\text{Zn}(1)—\text{S}(4)$ 71.28(2), $\text{N}(5)—\text{Zn}(1)—\text{N}(6)$ 75.00(9), $\text{S}(2)—\text{Zn}(1)—\text{S}(4)$ 168.49(3), $\text{S}(1)—\text{Zn}(1)—\text{N}(6)$ 166.79(7), $\text{S}(3)—\text{Zn}(1)—\text{N}(5)$ 155.07(6)°.

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evidence was also found for an alternate position of the C70 chloroform molecule by rotation about the C—H axis; attempts to resolve this disorder were unsuccessful. Programs used: teXsan, DIRDIF, SHELXL, and ORTEP. CCDC deposition number: 193904.

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